



برعاية معالي وزير التربية والتعليم

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وتوجيهات مساعد الوزير لشنون تطوير المناهج التطيمية والمشرف على الإدارة المركزية لتطوير المناهج

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الراءات وتقييمات

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Classroom performance



Lesson 1: Chemical reaction, Thermal Decomposition Reactions

Question 1: Write the Scientific Term:

- 1) The breaking bonds between the molecules of the reactants and forming new bonds between the molecules of the molecules resulting from the reaction.
- 2) Chemical reactions in which the thermal compound decomposes into its primary elements or simple compound.
- 3) A safety means is found in modern cars and has a form of an inflatable bag.

Question 2: Choose the correct answer:

When heating copper hydroxide, we get

(copper carbonate and water vapor - copper oxide and water vapor - copper and hydrogen - copper oxide and hydrogen)

2) Some metal nitrates decompose with heat into

(metal nitrite and oxygen - metal nitrate and oxygen - nitric oxide and hydrogen - metal nitrate)

3) When heating red mercury oxide, it decomposes into

(oxygen - mercury - oxygen and mercury - there is no correct answer)

4) Almost metal carbonates decomposed by heating into metals and

(O2 - NO2 - CO2 - SO3)

Question 3: What is the importance of the chemical reactions?

Question 4: What are the types of chemical reactions?

Question 5: according to your study

Mention the names of the chemical compounds that give oxygen gas when thermally decomposed.